

## Pyroelectric Energy Conversion Cycles Tailored for Antiferroelectrics

B.M. Hanrahan<sup>1\*</sup>, Y. Espinal<sup>1,2</sup>, C. J. Neville<sup>1</sup>, and A. N. Smith<sup>3</sup>

<sup>1</sup>U.S. Army Research Laboratory  
2800 Powder Mill Rd., Adelphi, MD 20783

<sup>2</sup>Institute of Materials Science, University of Connecticut  
97 North Eagleville Road, Storrs, CT 06269-3136

<sup>3</sup>U.S. Naval Academy  
121 Blake Rd, Annapolis, MD 21402

\*Brendan.M.Hanrahan.civ@mail.mil

Pyroelectric energy conversion research is accelerating towards realizing efficient, practical systems, which can be credited to two major innovations: electric field and temperature-driven thermodynamic cycles<sup>1</sup>, and thin-film active materials able to withstand high electric fields<sup>2</sup>. Antiferroelectrics have been recognized for their potential for energy conversion<sup>3</sup>, but there has not been a thermodynamic cycle analysis that takes advantage of the unique features of antiferroelectric polarization-temperature relationships. Specifically, the polarization at low electric fields increases as temperature increases, creating a region of inverse intrinsic pyroelectricity. This region is highlighted in figure 1. We have developed a two-part conversion cycle (figure 2) that takes advantage of this unique region while increasing energy density by over 17% for a given temperature range and enables cascaded heat transfer system designs. As materials with exotic polarizations are realized<sup>4</sup>, it will be important to re-examine their energy conversion cycles for specific property relationships.

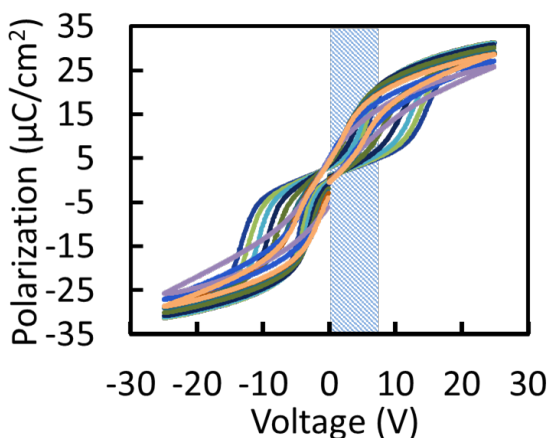


Figure 1. Isothermal hysteresis loops from 300-540K for a 500 nm  $(\text{Pb}_{0.995}\text{La}_{0.005})\text{Zr}_{(0.95)}\text{Ti}_{(0.05)}\text{O}_3$  thin film with Pt bottom and  $\text{IrO}_2$  top electrodes.

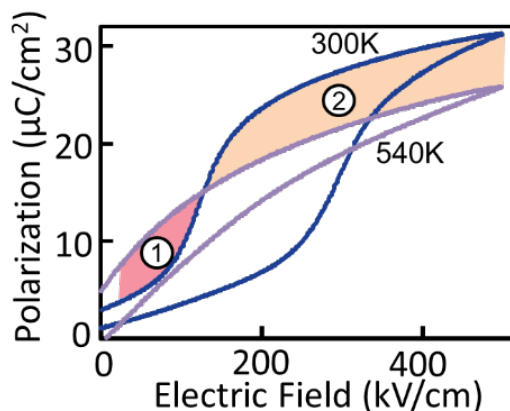


Figure 2. Energy conversion cycles between AFE and PE phases. Cycle 1 consists of heating at a low field while polarizing, charging, iso-polarization cooling, and discharging. Heat is transferred to cycle 2, which is an Olsen cycle [1].

1. R. B. Olsen, D. A. Bruno and J. M. Briscoe, *Journal of Applied Physics* **58** (12), 4709-4716 (1985).
2. A. S. Mischenko, Q. Zhang, J. F. Scott, R. W. Whatmore and N. D. Mathur, *Science* **311** (5765), 1270-1271 (2006).
3. X. Hao, Y. Zhao and Q. Zhang, *The Journal of Physical Chemistry C* **119** (33), 18877-18885 (2015).
4. J. Mangeri, K. C. Pitike, S. P. Alpay and S. Nakhmanson, *Npj Computational Materials* **2**, 16020 (2016).